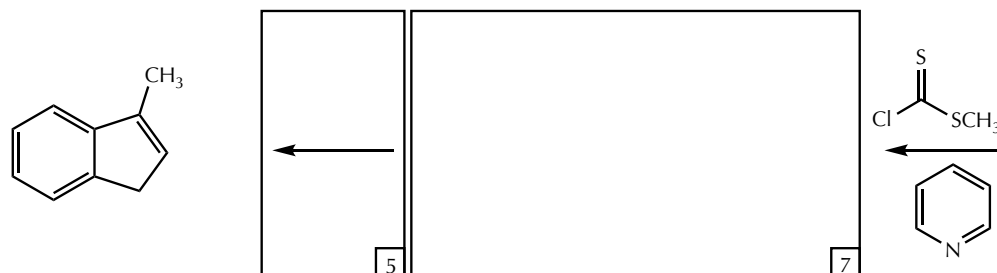
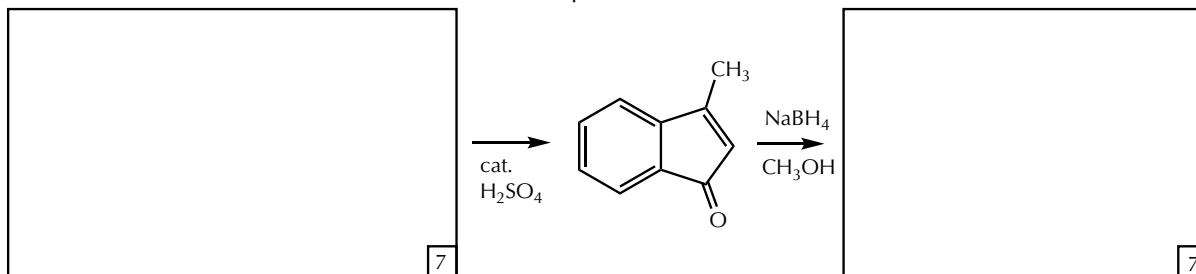


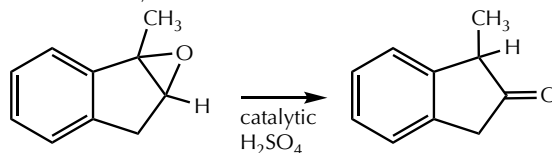
Question I (54 points)

Name: _____

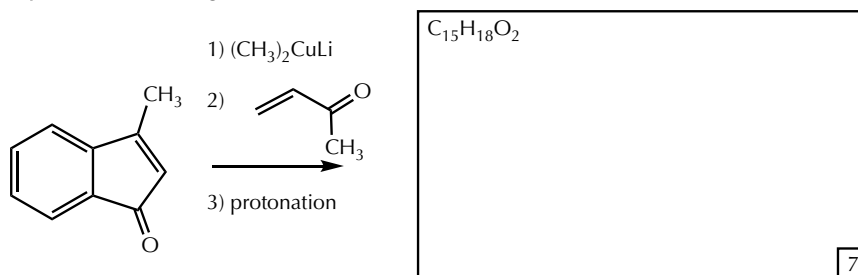
A. Complete the following reaction sequence (in part: *Org. Process Res. Dev.* **2022**, 26, 1960), which begins with an *intramolecular aldol condensation*. You do not need to provide stereochemical information.



B. When the epoxide derived from the product in part A, above, is treated with an acid catalyst, a rearrangement to a ketone is observed. Using $\text{HB}/\text{B}^\ominus$ as your generic Brønsted acid/base, as needed, provide the complete, curved arrow mechanism for this transformation, in which a carbocation intermediate is anticipated.



C. Complete the following transformation, which is carried out on the aldol condensation intermediate from part A.

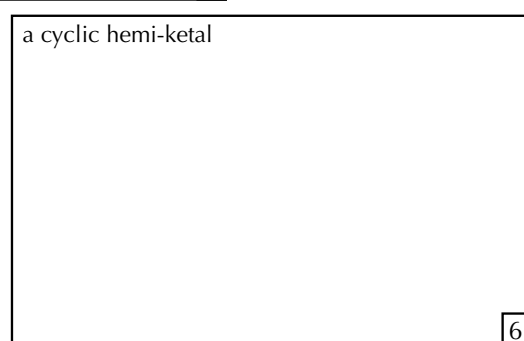
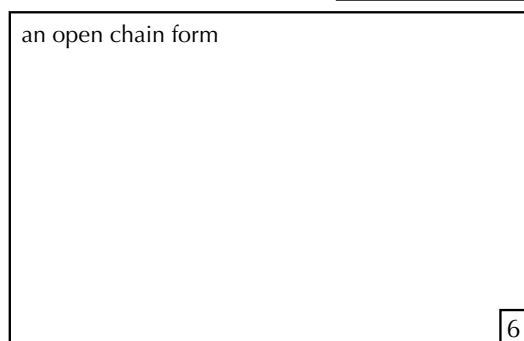
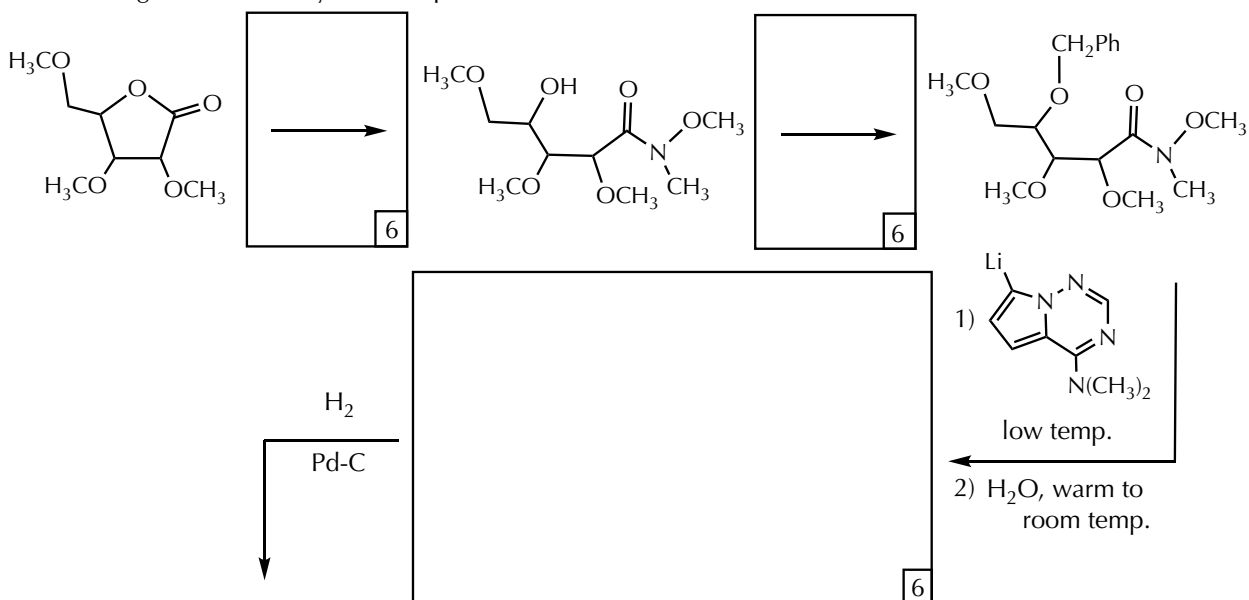


A	/	/	/26
B	/	/	/21
C	/	/	/07
	/	/	/54

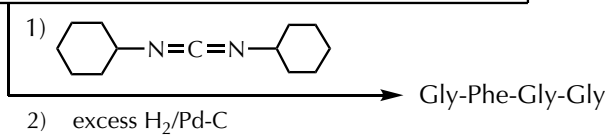
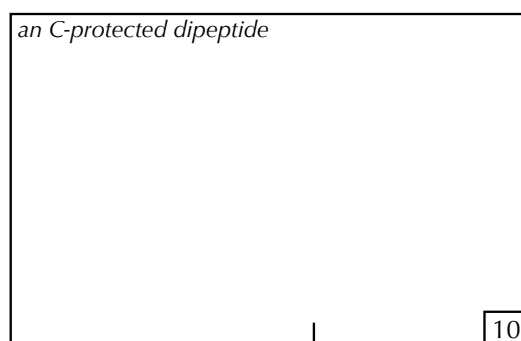
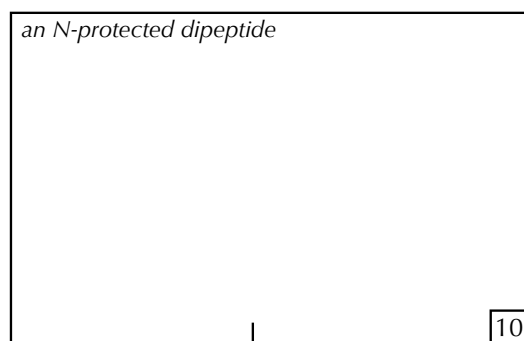
Question II (50 points)

Name: _____

- A. Complete the following reaction scheme that was used in a practical synthesis of remdesivir, the first and only FDA-approved antiviral drug for treating COVID-19 (*J. Org. Chem.* **2021**, *86*, 5065). Showing stereochemistry is not required.



- B. Taken from a synthesis of the osteogenic growth peptide (OGP), present in small concentrations in circulating blood (*Org. Process Res. Dev.* **2015**, *19*, 1257). **Show stereochemistry; no abbreviations.**



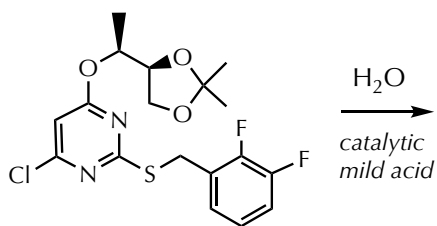
A	...	/30
B	...	/20
	...	/50

Question III (46 points)

Name: _____

Complete the following transformations.

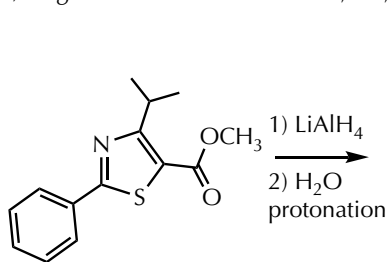
(a) *Org. Process Res. Dev.* **2022**, 26, 2337.



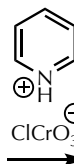
balance the equation; give the neutral/uncharged product(s)

6

(b) *Org. Process Res. Dev.* **2022**, 26, 10.



6



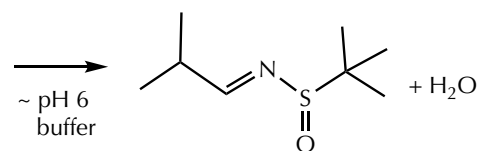
6

(c) *Org. Process Res. Dev.* **2022**, 26, 2138.

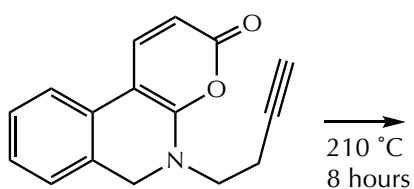
6

+

6



(d) *J. Org. Chem.* **1996**, 61, 1650.

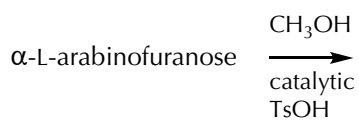


an intramolecular
Diels-Alder
reaction creates
an intermediate
that undergoes a
retro-Diels-Alder
reaction to lose
carbon dioxide

→

+ CO₂ 8

(e) arabinose is the C2 epimer of ribose



gives the β -anomer of the product

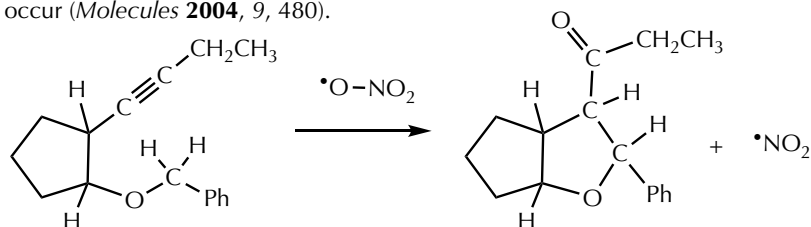
8

a	/	/06
b	/	/12
c	/	/12
d	/	/08
e	/	/08
	/	/46

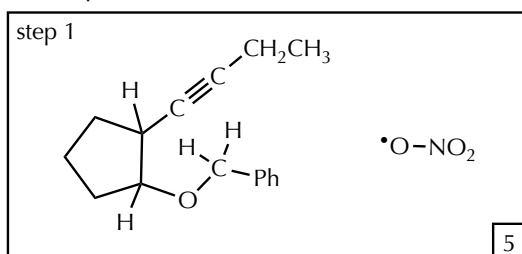
Question IV (48 points)

Name: _____

- A. The photochemical decomposition of $(\text{NH}_4)_2 [\text{Ce}(\text{NO}_3)_6]$ gives an oxygen atom radical: $\bullet\text{O}-\text{NO}_2$. The following reaction is observed to occur (*Molecules* **2004**, 9, 480).



The mechanistic steps are outlined here: provide the missing intermediates as well as the curved (fish-hook) arrows for each step.



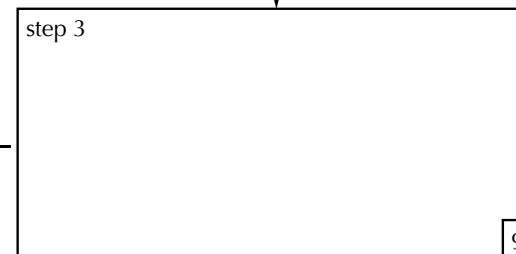
the oxygen atom adds to the triple bond; its regioselectivity can be inferred from the product



the sp^2 carbon radical resulting from step 1 removes a hydrogen atom intramolecularly from the benzyl group



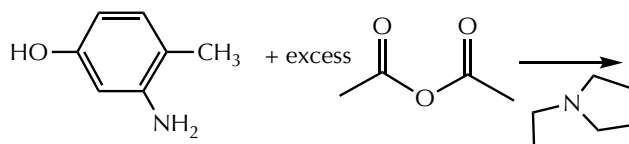
the oxygen-stabilized radical formed in step 3 gives a dissociation reaction resulting in the observed products (drawn above)



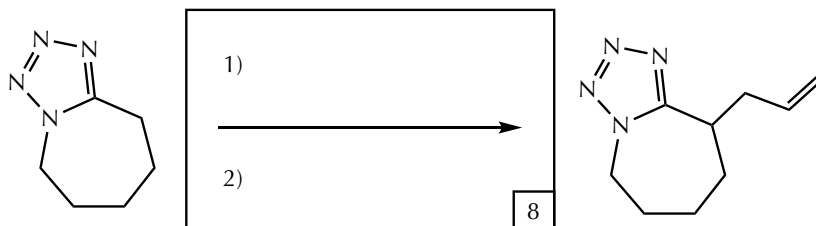
the benzylic carbon radical formed in step 2 undergoes an intramolecular addition reaction to give an oxygen-stabilized radical

- B. Complete the following transformations.

- (a) *Org. Process Res. Dev.* **2022**, 26, 10.



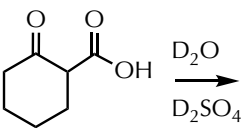
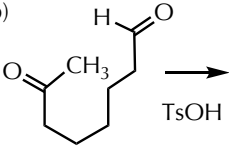
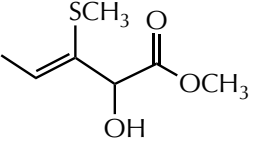
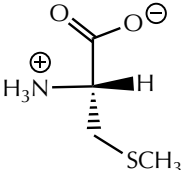
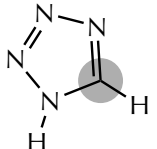
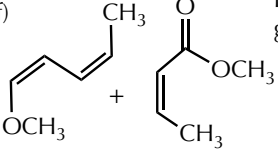
- (b) *Org. Lett.* **2022**, 24, 6722.

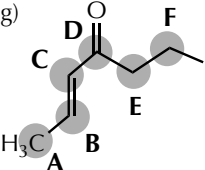
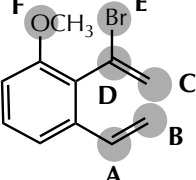
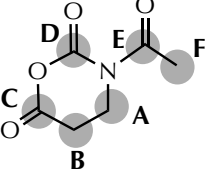
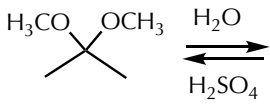


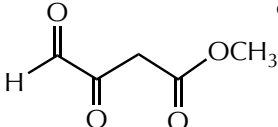
A	...	/32
B	...	/16
	...	/48

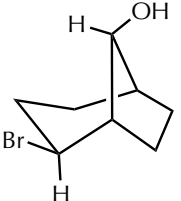
Question V (42 points)

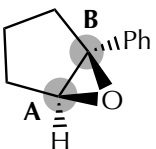
Name: _____

- (a)  How many exchangeable proton under H/D exchange conditions?
- (b)  Number of possible intramolecular aldol condensations?
- (c)  Position number of the alcohol group in the IUPAC name?
- (d)  Stereochemical configuration [(R) or (S)] for methionine?
- (e)  Oxidation number of the shaded atom?
- (f)  Relationship of the 2 CH₃ groups in the major product? *circle one*

1,2-cis	1,2-trans
1,3-cis	1,3-trans
- (g)  Site (A-F) of reactivity of this compound with an enol under acid conditions?
- (h)  Site (A-F) of fastest reactivity with tributyltin radical?
- (i)  Most reactive site (A-F) with methyl magnesium bromide?
- (j)  Based on the entropy change, the K_{EQ} for this process is: *circle one*

> 1	~ 1	< 1
can't tell		
- (k) L-mannose is the open chain form of an aldohexose. How many chiral diastereomers does L-mannose have?
- (l)  The IUPAC name for this compound would include: *circle one*

trioxo	dioxo	oxo
no "oxo"		
- (m)  The position of the two groups, "OH" & "Br", are: *circle one*

exo & exo	exo & endo
endo & exo	endo & endo
- (n)  Site of fastest reaction with ammonia (NH₃)?

number of correct answers:

1	2	3	4	5	6	7	8	9	10	11	12	13	14
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 score:

3	6	9	12	15	18	21	24	27	30	33	36	39	42
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